

Two Distinct Mechanisms for TIM Barrel Prenyltransferases in Bacteria

Emma H. Doud,^{†,‡} Deborah L. Perlstein,[‡] Manuel Wolpert,[‡] David E. Cane,[§] and Suzanne Walker^{*,†}

[†]Department of Chemistry and Chemical Biology, Harvard University, 12 Oxford Street, Cambridge, Massachusetts 02139, United States

[‡]Department of Microbiology and Molecular Genetics, Harvard Medical School, 200 Longwood Avenue, Boston, Massachusetts 02115, United States

⁸Department of Chemistry, Brown University, Box H, Providence, Rhode Island 02912-9108, United States

S Supporting Information

ABSTRACT: The reactions of two bacterial TIM barrel prenyltransferases (PTs), MoeO5 and PcrB, were explored. MoeO5, the enzyme responsible for the first step in moenomycin biosynthesis, catalyzes the transfer of farnesyl to 3-phosphoglyceric acid (3PG) to give a product containing a cis-allylic double bond. We show that this reaction involves isomerization to a nerolidyl pyrophosphate intermediate followed by bond rotation prior to attack by the nucleophile. This mechanism is unprecedented for a prenyltransferase that catalyzes an intermolecular coupling. We also show that PcrB transfers geranyl and geranylgeranyl groups to glycerol-1-phosphate (G1P), making it the first known bacterial enzyme to use G1P as a substrate. Unlike MoeO5, PcrB catalyzes prenyl transfer without isomerization to give products that retain the trans-allylic bond of the prenyl donors. The TIM barrel family of PTs is unique in including enzymes that catalyze prenyl transfer by distinctly different reaction mechanisms.

Tature uses prenyltransferases (PTs) to make a wide variety N of primary and secondary metabolites containing prenyl groups, and there is enormous interest in understanding the scope and utility of these enzymes.¹ The best studied classes of PTs, the polyprenyl synthases and the cyclizing terpene synthases, contain an alpha helical fold, as do the protein farnesyl and geranylgeranyl transferases.² Recently, the first PT having a triosephosphate isomerase (TIM) barrel fold was identified in archaea.³⁻⁵ This PT, geranylgeranylglyceryl phosphate synthase (GGGPS), catalyzes the transfer of a geranylgeranyl (C20) group to glycerol-1-phosphate (G1P) during the first step in the biosynthesis of the major phospholipid component of archaeal membranes (Figure 1).^{6,7} We recently discovered a bacterial GGGPS homologue, MoeO5, in the gene cluster for the antibiotic moenomycin. MoeO5 couples 3-phosphoglyceric acid (3PG) to a farnesyl group to form the starter unit for moenomycin biosynthesis (Figure 1).^{8,9} Here, we study the mechanism of MoeO5, describe key residues that can be used to predict the acceptor substrate in TIM barrel PTs, apply these criteria to a bacterial GGGPS homologue of unknown function, PcrB, and establish its substrates biochemically. We conclude that TIM barrel PTs are dedicated to the transfer of prenyl groups to

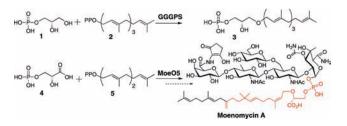


Figure 1. GGGPS converts glycerol-1-phosphate (G1P, 1) and GGPP (2) to compound 3. MoeO5 farnesylates 3-phosphoglycerate (3PG, 4) using farnesyl pyrophosphate (FPP, 5). The portion of MmA synthesized by MoeO5 is highlighted in red.

oxygen on phosphorylated triose acceptors, but the reactions they catalyze can proceed by two very different mechanisms.⁴

The TIM barrel PTs, GGGPS and MoeO5, both make etherlinked products but with different allylic bond geometries (Figure 1). Unlike the GGGPS product, which retains the trans-geometry of the starting material and likely proceeds by a straightforward displacement mechanism, the MoeO5 product (6) contains a *cis*-allylic double bond.⁸ MoeO5 is the only known prenyltransferase that forms an intermolecular cis-allylic bond using a *trans*-allylic precursor.¹⁰ This linkage suggests that bond isomerization occurs prior to coupling during the catalytic reaction, and a possible reaction mechanism involving the intermediacy of nerolidyl pyrophosphate (NPP, 7) is shown in Figure 2A.¹

To test this hypothesis, we compared the reaction rates for FPP (5) and (3S)-NPP (7) using LC/MS. Under initial rate conditions, the turnover numbers for FPP and NPP were indistinguishable (7.8 \pm 0.1 \min^{-1} for FPP and 7.4 \pm 0.7 \min^{-1} for NPP).¹² In addition, we detected a compound with a retention time and mass consistent with those of NPP (7) in reactions of MoeO5 incubated with FPP in the absence of 4 (Figure 2B).¹³ These results support a mechanism in which MoeO5 isomerizes FPP (5) to NPP (7), which undergoes rotation about the C-2,3-bond to the cisoid conformer prior to prenyl transfer to phosphoglyceric acid (4, Figure 2).

To gain insight into how MoeO5 and GGGPS are capable of catalyzing prenyl transfer to acceptors in different oxidation states via fundamentally different mechanisms, we aligned the sequences of 50 archaeal TIM barrel PTs and MoeO5 and

Received: November 3, 2010 Published: January 7, 2011

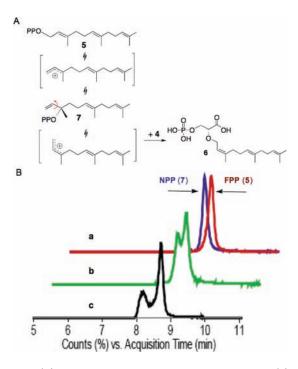


Figure 2. (A) MoeOS catalyzes the formation of 6 from 3PG (4) and FPP (5) via a mechanism proposed to involve isomerization to NPP (7). (B) NPP forms during the MoeOS reaction. NPP and FPP have the same $[M - H]^-$ (381.1237). LC/MS extracted ion chromatograms of an (a) overlay of NPP and FPP standards (blue and red, respectively) demonstrating reproducible differences in retention times; (b) MoeOS reaction of FPP spiked with NPP following quenching of the enzymatic reaction (green); (c) MoeOS reaction with FPP showing the formation of a new peak with an RT consistent with NPP (black).¹⁴

A	200 210	
MoeO5	HMVYLYSRN-EHVPPEV	/VR
Afgggps	PIIYIEYSGTYGNPE	LV
HbGGGPS	EVVYVEYSGTLGDPA	A V V
FpGGGPS	PIIYVEYSGTYGDPE	LV
Tagggps	RLVYFEAGSGSPYHVGE	ΣNV
MtGGGPS	RLFYLEAGSGAPEHVPE	ЕМ
PhGGGPS	RLVYLEAGSGAPEPVPE	ЕМ
BsPcrb	PIFYLEYSGVLGDIE	ΕAV
в	100	109
B MoeO5	100 P L P V V L <mark>H F P P R P</mark> G A G F E	
		vv
MoeO5	P L P V V L <mark>H F P P R P</mark> G A G F E	v v v v
MoeO5 AfGGGPS	P L P V V L <mark>HF P P R P</mark> G A G F E - L P I V V E P S D P S N	V V I V V I V E
MoeO5 AfGGGPS HbGGGPS	P L P V V L <mark>H F P P R P</mark> G A G F F - L P I V V E P S D P S N - V P L Y Q E P S N P A V	V V V V V V V V V V V V
MoeO5 AfGGGPS HbGGGPS FpGGGPS	P L P V V L <mark>H F P P R P</mark> G A G F E - L P I V V E P S D P S N - V P L Y Q E P S N P A V D K P I I M E P S D P S N	VV VV VE VV UV US
MoeO5 AfGGGPS HbGGGPS FpGGGPS TaGGGPS	P L P V V L H F P P R P G A G F E - L P I V V E P S D P S N - V P L Y Q E P S N P A V D K P I I M E P S D P S N D L K V I L F P G S S L M	VV IVV IVV IVV IIS
MoeO5 AfGGGPS HbGGGPS FpGGGPS TaGGGPS MtGGGPS	P L P V V L H F P P R P G A G F E - L P I V V E P S D P S N - V P L Y Q E P S N P A V D K P I I M E P S D P S N D L K V I L F P G S S L M D V P I I L F P G N T T	VV VE VV IVV IS VS

Figure 3. Partial sequence alignments of MoeO5 (*S. ghanaensis*), representative archaeal GGGPSs, and PcrB from *B. subtilis* (BsPcrB). (A) Alignments of the acceptor binding loop. The tyrosine and arginine changes in MoeO5 are outlined in red. (B) Alignments of the "swinging door" region of AfGGGPS described in the text.

compared them to the GGGPS crystal structure, which contains a bound acceptor substrate.³ Amino acids in the β -6-loop-6 region of GGGPS make direct contacts to the acceptor substrate, and there are several striking changes in this region of MoeO5 (Figure 3).³ For example, E167, which contacts both the C-2 and C-3 hydroxyl groups of G1P in the GGGPS crystal structure, is replaced by tyrosine. This change likely prevents repulsive interactions with the 3PG carboxylate. In addition, S169 is

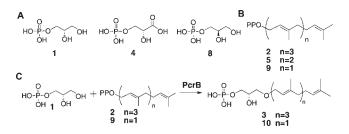


Figure 4. (A) Phosphosugar acceptors tested as substrates for PcrB: G1P (1), 3PG (4), and G3P (8). (B) Prenylpyrophosphate donors tested as substrates for PcrB: GGPP (2), FPP (5), and GPP (9). (C) The reactions shown to be catalyzed by PcrB.

replaced by arginine, which may favor attractive interactions with the substrate carboxylate.

Changes in the acceptor binding loop help explain the changes in acceptor substrate preferences, but other features are likely responsible for the ability of MoeO5 to catalyze isomerization of the allylic diphosphate bond of FPP. Notably, MoeO5 contains a proline-rich insertion (Figure 3B) between β -3 and α -3*, a region in GGGPS that is suggested to form a "swinging door" that closes over the prenyl chain.³ In addition, while the GGGPS binding tunnel is largely hydrophobic and acidic, the MoeO5 homology model shows a nest of basic and π -electron containing residues in the vicinity of the "swinging door" that could help stabilize a partially ionized nerolidyl intermediate. Using the absence of glutamate and the presence of arginine in the acceptor binding loop, combined with the presence of a proline insertion in the α -3^{*} strand, we have identified several MoeO5 homologues in other bacteria, always clustered with genes putatively involved in secondary metabolite biosynthesis.1

We also investigated the enzymatic function of PcrB, a putative prenyltransferase with a TIM barrel fold.¹⁶ PcrB is conserved in several Gram-positive genera, including *bacilli*, *staphylococci*, and *listeriae*. Based on sequence analysis, it has been speculated that PcrB either is an aromatic prenyltransferase or has a function similar to GGGPS.^{37,17} Since PcrB contains an acceptor binding loop and "swinging door" similar to those in GGGPS, it seemed likely that PcrB is a eubacterial prenyltransferase that uses G1P as a substrate.

To test this possibility, we cloned, expressed, and purified Bacillus subtilis PcrB and incubated it with three potential acceptors, glycerol-1-phosphate (G1P, 1), glycerol-3-phosphate (G3P, 8), and 3-phosphoglycerate (3PG, 4), and three potential prenyl donors, GGPP (2), FPP (5), and GPP (9) (Figure 4A and B).¹⁴ The reactions, analyzed by mass spectrometry, showed that PcrB uses G1P (1) exclusively as an acceptor but is able to transfer both the C10 and C20 carbon donors (GPP, 9; GGPP, 2) at similar rates (Figure 3C, Table S1, turnover: 1.8 \pm 0.1 and $0.7 \pm 0.1 \text{ min}^{-1}$, respectively); furthermore, upon extended incubation both reactions went to completion.¹⁸ In contrast, FPP (5), which contains 15 carbons, is a poor substrate for PcrB, and reactions using it as a donor proceeded to less than 10% conversion. NMR analysis showed that the PcrB product formed using GPP as a donor has structure 10 (Figure 3, Figures S1-S3) in which the allylic bond is *trans* and the ether linkage is to the C3 hydroxyl of G1P.14 Hence the reaction is identical to that catalyzed by GGGPS, except that both short (C10) and long (C20) prenyl chains are substrates.¹⁹ Although G1P (1) is a common building block for membrane lipids in archaea, PcrB is

the first known bacterial enzyme to use it as a substrate.^{17,20} The *pcrB* gene is always found in the same operon as *pcrA*,²¹ which encodes an essential helicase involved in the replication and transfer of DNA elements between bacteria.²² This genetic linkage suggests a possible functional connection. Knowledge of PcrB's enzymatic function may facilitate efforts to elucidate its cellular role.

The reactions catalyzed by GGGPS, MoeOS, and PcrB suggest that the TIM barrel family of PTs is dedicated to the transfer of prenyl groups to oxygen acceptors of triose phosphates. We have shown that the oxidation state of the acceptor and the regiochemistry of reaction can vary depending on the PT. More remarkably, this family of enzymes can catalyze the formation of products containing either *cis- or trans*-allylic double bonds from *trans*-prenyl donors.^{14,23} Thus, relatively small changes in three-dimensional architecture lead to prenyl transfer by two distinct mechanisms, one involving direct displacement and the other requiring bond isomerization and rotation. This is unprecedented for a single PT family. Structural information on MoeO5 or its homologues should provide clues to the specific sequence changes that have led to this dramatic switch in mechanism.

ASSOCIATED CONTENT

Supporting Information. Experimental Procedures, including cloning and purification of proteins, and raw MS and NMR data, complete refs 16 and 21b. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author Suzanne walker@hms.harvard.edu

ACKNOWLEDGMENT

We thank Charles Sheahan for training and assistance at the HMS East Quad NMR Facility. This work was partially supported by NERCE (NIAID U54AI057159) as well as NIH Grants GM076710 and AI083214 (S.W.) and GM30301 (D.E.C.). E.H.D. was supported by an NSF graduate research fellowship, and D.L.P. was supported by an NIH postdoctoral fellowship. All LC/MS data were acquired on an Agilent 6520 Q-TOF spectrophotometer supported by the Taplin Funds for Discovery Program (P.I.: S.W.). We thank the Biomolecule Production Core of NERCE for reagents (NIAID U54AI057159).

REFERENCES

(1) (a) Dewick, P. M. Nat. Prod. Rep. 2002, 19, 181–222. Heide, L. Curr. Opin. Chem. Biol. 2009, 13, 171–179. (b) Liang, P. H. Biochemistry 2009, 48, 6562–6570.

(2) (a) Liang, P. H.; Ko, T. P.; Wang, A. H. J. Eur. J. Biochem. 2002, 269, 3339–3354. (b) Christianson, D. W. Chem. Rev. 2006, 106, 3412–3442. (c) Greenhagen, B. T.; O'Maille, P. E.; Noel, J. P.; Chappell, J. Proc. Natl. Acad. Sci. U.S.A. 2006, 103, 9826–9831. (d) Lane, K. T.; Beese, L. S. J. Lipid Res. 2006, 47, 681–699. (e) Thulasiram, H. V.; Erickson, H. K.; Poulter, C. D. Science 2007, 316, 73–76. (f) Vedula, L. S.; Zhao, Y. X.; Coates, R. M.; Koyama, T.; Cane, D. E.; Christianson, D. W. Arch. Biochem. Biophys. 2007, 466, 260–266. (g) Shishova, E. Y.; Yu, F. L.; Miller, D. J.; Faraldos, J. A.; Zhao, Y. X.; Coates, R. M.; Allemann, R. K.; Cane, D. E.; Christianson, D. W. J. Biol. Chem. 2008, 283, 15431–15439. (h) Aaron, J. A.; Lin, X.; Cane, D. E.; Christianson, D. W. Biochemistry 2010, 49, 1787–1797.

(3) Payandeh, J.; Fujihashi, M.; Gillon, W.; Pai, E. F. J. Biol. Chem. 2006, 281, 6070–6078.

(4) TIM barrel proteins catalyze a wide range of enzymatic reactions; their structural details, mechanism, and evolution have been the focus of numerous studies. See: (a) Reardon, D.; Farber, G. K. FASEB J. **1995**, *9*, 497–503. (b) Copley, R. R.; Bork, P. J. Mol. Biol. **2000**, 303, 627–640. Lang, D.; Thoma, R.; Henn-Sax, M.; Sterner, R.; Wilmanns, M. Science **2000**, 289, 1546–1550. (c) Wierenga, R. K. FEBS Lett. **2001**, 492, 193–198. (d) Nagano, N.; Orengo, C. A.; Thornton, J. M. J. Mol. Biol. **2002**, 321, 741–765. (e) Anantharaman, V.; Aravind, L.; Koonin, E. V. Curr. Opin. Chem. Biol. **2003**, 7, 12–20. (f) Gerlt, J. A.; Raushel, F. M. Curr. Opin. Chem. Biol. **2003**, 7, 252–264.

(5) A distinct PT barrel structure has recently been determined for several aromatic prenyltransferases, including Orf2 and CloQ. See: (a) Tello, M.; Kuzuyama, T.; Heide, L.; Noel, J. P.; Richard, S. B. *Cell. Mol. Life Sci.* **2008**, *65*, 1459–1463. (b) Kuzuyama, T.; Noel, J. P.; Richard, S. B. *Nature* **2005**, *435*, 983–987.

(6) (a) Chen, A. J.; Zhang, D. L.; Poulter, C. D. J. Biol. Chem. 1993, 268, 21701–21705. (b) Zhang, D. L.; Poulter, C. D. J. Am. Chem. Soc. 1993, 115, 1270–1277. (c) Zhang, D. L.; Poulter, C. D. J. Org. Chem. 1993, 58, 3919–3922. (d) Soderberg, T.; Chen, A. J.; Poulter, C. D. Biochemistry 2001, 40, 14847–14854.

(7) Nemoto, N.; Oshima, T.; Yamagishi, A. J. Biochem. 2003, 133, 651-657.

(8) Ostash, B.; Doud, E. H.; Lin, C.; Ostash, I.; Perlstein, D. L.; Fuse, S.; Wolpert, M.; Kahne, D.; Walker, S. *Biochemistry* **2009**, *48*, 8830–8841.

(9) Ostash, B.; Saghatelian, A.; Walker, S. Chem. Biol. 2007, 14, 257-267.

(10) The *cis*-geometry observed in the products of *cis*-prenyl chain elongating enzymes results from a different type of reaction mechanism. See: Kharel, Y.; Koyama, T. *Nat. Prod. Rep.* **2003**, *20*, 111–118.

(11) Nerolidyl diphosphate is a demonstrated intermediate in the cyclization—isomerization—cyclization of FPP to 6-, 10-, and 11-membered ring sesquiterpenes with *cis*-double bonds. See: (a) Cane, D. E.; Ha, H. J.;Pargellis, C.; Waldmeier, F.; Swanson, S.; Murthy, P. P. N. *Bioorg. Chem.* **1985**, *13*, 246–265. (b) Cane, D. E.; Ha, H. J. *J. Am. Chem. Soc.* **1988**, *110*, 6865–6870. (c) Cane, D. E.; Pawlak, J. L.; Horak, R. M.; Hohn, T. M. *Biochemistry* **1990**, *29*, 5476–5490.(d) Domingo, V.; Arteaga, J. F.; del Moral, J. F. Q.;(e) Barrero, A. F. *Nat. Prod. Rep.* **2009**, *26*, 115–134. (f) Noel, J. P.; Dellas, N.; Faraldos, J. A.; Zhao, M.; Hess, B. A.; Smentek, L.; Coates, R. M.; O'Maille, P. E. *ACS Chem. Biol.* **2010**, *5*, 377–392.

(12) Rates were measured in triplicate using 1 mM 3PG, 40 μ M lipid pyrophosphate, and 120 nM enzyme. Conversion was <10%. See Supporting Information (SI) for full details.

(13) We cannot rule out that the intermediate detected could also be *cis,trans*-farnesylpyrophosphate.

(14) See SI.

(15) MoeOS homologues are present and clustered with moenomycin type biosynthetic clusters in *S. clavuligerus, Photohrabdus luminescens subsp. laumondii* TTO1, and *P. asymbiotica*.

(16) Badger, J.; et al. Proteins: Struct., Funct., Bioinf. 2005, 60, 787–796.
(17) Guldan, H.; Sterner, R.; Babinger, P. Biochemistry 2008, 47, 7376–7384.

(18) Rates were measured in triplicate using 1 mM G1P, 40 μ M lipid pyrophosphate, and 3.4 μ M enzyme. Conversion was <15%. See SI for full details.

(19) How FPP is excluded is unknown, although we note that the hydrophobic tunnel in PcrB has a kink that would require substrates longer than 10 carbons to bend.

(20) A bacterial enzyme that interconverts glyceraldehyde-phosphate and G1P was recently identified, which suggests that bacteria produce this metabolite for some purpose. See: ref 17.

(21) (a) Ji, Y. D.; Zhang, B.; Van Horn, S. F.; Warren, P.; Woodnutt,
G.; Burnham, M. K. R.; Rosenberg, M. Science 2001, 293, 2266–2269.
(b) Kobayashi, K.; et al. Proc. Natl. Acad. Sci. U.S.A. 2003, 100, 4678–4683.

Journal of the American Chemical Society

(22) (a) Ruiz-Maso, J. A.; Anand, S. P.; Espinosa, M.; Khan, S. A.; del Solar, G. *J. Bacteriol.* **2006**, *188*, 7416–7425. (b) Slatter, A. F.; Thomas, C. D.; Webb, M. R. *Biochemistry* **2009**, *48*, 6326–6334. (c) Lee, C. A.; Babic, A.; Grossman, A. D. *Mol. Microbiol.* **2010**, *75*, 268–279.

(23) Using homology searches, we have identified a putative new subfamily of TIM barrel PTs that are highly conserved in Gram-negative bacteroides, and we predict that these enzymes also use G1P as a substrate and produce a *trans* product (Figure S4).